

Chapter 9 Stoichiometry Test Answers

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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g Calculate the percentage yield 2 60 mol of N₂ are mixed with 120 mol of H₂ according to the following equation: N₂(g) + 3H₂(g)

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Download File PDF Chapter 9 Test Stoichiometry Answers Chapter 9 Test Stoichiometry Answers CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided 1 Given the following equation: C₃H₄(g) + xO₂(g) → 3CO₂(g) + 2H₂O(g) 4 a What is the value of the coefficient x in this equation? 40

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Modern Chemistry Chapter 9 Mixed Review Stoichiometry Answers CHAPTER 9 REVIEW Stoichiometry SECTION 92 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 1 The following equation represents a laboratory preparation for oxygen gas: CHAPTER 9 REVIEW Chapter 9 Review Stoichiometry Modern Chemistry

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